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How to cite:
Costello, Lacey J.; Filiberto, Justin; Crandall, Jake R.; Potter-McIntyre, Sally L.; Schwenzer, Susanne P.; Miller, Michael A.; Hummer, Daniel R.; Olsson-Francis, Karen and Perl, Scott (2020). Habitability of hydrothermal systems at Jezero and Gusev Craters as constrained by hydrothermal alteration of a terrestrial mafic dike. Geochemistry, 80(2), article no. 125613.

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Version: Version of Record

Link(s) to article on publisher’s website:
http://dx.doi.org/doi:10.1016/j.chemer.2020.125613

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Habitability of hydrothermal systems at Jezero and Gusev Craters as constrained by hydrothermal alteration of a terrestrial mafic dike

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ARTICLE INFO

Handling Editor: Astrid Holzheid
Keywords:
Hydrothermal
Mars alteration
Terrestrial analog
Gusev Crater
Jezero Crater
NE Syrtis

ABSTRACT

NASA’s search for habitable environments has focused on alteration mineralogy of the Martian crust and the formation of hydrous minerals, because they reveal information about the fluid and environmental conditions from which they precipitated. Extensive work has focused on the formation of alteration minerals at low temperatures, with limited work investigating metamorphic or high-temperature alteration. We have investigated such a site as an analog for Mars: a mafic dike on the Colorado Plateau that was hydrothermally altered from contact with groundwater as it was emplaced in the porous and permeable Jurassic Entrada sandstone. Our results show evidence for fluid mobility removing Si and K but adding S, Fe, Ca, and possibly Mg to the system as alteration progresses. Mineralogically, all samples contain calcite, hematite, and kaolinite; with most samples containing minor anatase, barite, halite, and dolomite. The number of alteration minerals increase with alteration. The hydrothermal system that formed during interaction of the magma (heat source) and groundwater would have been a habitable environment once the system cooled below ∼120 °C. The mineral assemblage is similar to alteration minerals seen within the Martian crust from orbit, including those at Gusev and Jezero Craters. Therefore, based on our findings, and extrapolating them to the Martian crust, these sites may represent habitable environments which would call for further exploration and sample return of such hydrothermally altered igneous materials.

1. Introduction

There is extensive evidence for water activity on ancient Mars from orbital investigations of geomorphic features (fluvial valleys, outflow channels, fans, deltas, paleolakes, lobate craters) (Carr, 1996; Carr and Head, 2010; Lasue et al., 2019) and in situ rover investigations of ancient sedimentary rocks and structures (conglomerates and mudstones, grain size sorting, cross bedding, and rounding of pebbles) (McLennan, 2012; Williams et al., 2013). Diverse alteration minerals have been discovered in the Martian crust from orbit (Ehlmann and Edwards, 2014), in Martian meteorites (Treiman et al., 1993; Bridges et al., 2001, 2019), as well as in situ from rover investigations (Arendt and Catalano, 2018; Jolliff et al., 2019; Mittlefehldt et al., 2019; Sutter et al., 2019). Clay minerals are predominantly found in the ancient Noachian and possibly into the early Hesperian crust preserving evidence of weathering, hydrothermal activity, and diagenesis (Mustard et al., 2008; Ehlmann and Edwards, 2014). Localized carbonates and chlorides have been found within paleolake deposits (Ehlmann et al., 2008) including the deltaic deposit in Jezero Crater (Goudge et al., 2015; Salvatore et al., 2018), which was chosen as the landing site for the NASA Mars 2020 rover. Possible evidence for hydrothermal activity in the form of silica, hematite, and sulfate sand, as well as localized carbonate deposits, were also investigated by MER Spirit at Gusev Crater (Morris et al., 2010; Filiberto and Schwenzer, 2013; Ruff and
Farmer, 2016). Furthermore, two groups of Martian meteorites, the nakhlites and orthopyroxenite ALH84001, contain carbonates that record evidence of alteration (Bridge et al., 2001; Bridges and Schwenzer, 2012; Bridges et al., 2019). At more recent times in Martian history a predominance of sulfates and iron-oxides in crustal deposits analyzed from orbit (Bibring et al., 2006; Ehlmann and Edwards, 2014) suggests a drying out of the surficial part of the crust.

The Martian crust is primarily basaltic in composition (McSween et al., 2003; Ehlmann and Edwards, 2014; Filiberto, 2017), thus the observed alteration minerals have formed from a mafic protolith in a variety of conditions (pressure, temperature, and fluid composition). To understand reaction pathways, terrestrial analogue studies for basalt alteration have largely focused on palagonite formation in Hawaii and Iceland (Morris et al., 2001; Bishop et al., 2002). Palagonite typically forms by alteration and hydration of basaltic glass at high water to rock ratios to form an alteration gel or fibrous material with clays, zeolites, oxides and carbonates precipitating in pore space (Stroncik and Schmincke, 2002). However, both sites (Hawaii and Iceland) are dominated by alteration of mafic glasses through interaction with seawater (Staudigel and Hart, 1983; Drief and Schiffman, 2004), which is not a process found to date on Mars (Bain and Margulis, 1983). Further, extensive work has focused on either modeling or experimentally investigating low temperature alteration (e.g., Griffith and Shock, 1995, 1997; Zolotov and Mironenko, 2007; McAdam et al., 2008; Schwenzer and Kring, 2009; Marion et al., 2011; Filiberto and Schwenzer, 2013; Schwenzer and Kring, 2013; Melwani Daswani et al., 2016; Zolotov and Mironenko, 2016; Hurowitz et al., 2006; Tosca and McLennan, 2009), with fewer studies focusing on higher temperature metamorphic conditions or magmatic hydrothermal systems (e.g., Schulze-Makuch et al., 2007; McCubbin et al., 2009; Filiberto et al., 2014; McCanta et al., 2014; McSween, 2015; Ruff and Farmer, 2016; Semprich et al., 2019).

Therefore, we have investigated the mineral assemblage and bulk chemistry changes that occur in situ hydrothermal alteration of a mafic dike with groundwater on the Colorado Plateau. Our results allow us to predict what constraints this interaction can place on habitability of this environment as well as similar environments on Mars – specifically at Gusev Crater and the regional area around Jezero Crater in North East Syrtis.

1.1. Geologic background

On and near the Colorado Plateau in south-central Utah, hundreds of mafic dikes were emplaced into Jurassic sedimentary rocks, at the eastern limit of the Tertiary volcanic field (Delaney and Gartner, 1997). Our investigation focuses on the 22 Ma Robbers Roost dike, located at 38°30′58.52″N, 110°26′32.57″W, which is exposed several kilometers east of the main cluster of dikes (Wannamaker et al., 2000). The dike is a light rare earth element enriched olivine-phlogopite-lamproite that intruded pre-existing northwest oriented fracture systems in the crust (Wannamaker et al., 2000). The dike intruded through the Jurassic Entrada Sandstone of the San Rafael Group, an iron-cemented red silty-sandstone (Crabaugh and Kocurek, 1993). The Entrada varies regionally between siltstone and sandstone of various proportions, and exposures to the West of the Green River differ significantly from exposures east of the river. In the San Rafael Swell, the Entrada Sandstone is mostly very fine grained sandstone to siltstone with four partly crossbedded sandstone units (Wright et al., 1979; O’ Sullivan, 1981).

1.2. Field description

The dike can be separated into four visually distinct zones based on differences in colors and textures presumably correlated to different degrees of alteration (Fig. 1). The region of the intrusion located at the far northwest limit of the exposure is the freshest in appearance and darkest in color. The next distinct zone is yellow-green in coloration and is overall more friable than the other three sections. Continuing to the southeast, the intrusion is a deep purple-red color. The final zone of the dike lies at the southeastern extent of the exposed intrusion and is of bright crimson color. The surrounding sandstone is preferentially harder than the dike with the darkest zone of the dike being the most consolidated and resistant to weathering. The purple and red zones of the dike are increasingly less weathering resistant and thus of rubbly and weathered appearance. The green zone is the most rubbly in appearance, represents the smallest portion of the exposure, and the outcrop itself is the least continuous.

2. Methods

For bulk chemistry, one representative aliquot from each section of the intrusion (dark, green, purple, and red; Fig. 2) was powdered and sent to the Southwest Research Institute in San Antonio (SWRI*). The bulk powder was analyzed semi-quantitatively using an XRF Systems (Model 550i) energy dispersive X-ray spectrometer (EDS) attached to a Philips (Model XL-40) scanning electron microscope. Samples were analyzed in scanning mode using low magnification (100-300x) and a ~1 μm beam to ensure that the relative elemental abundances determined from the integrated X-ray detector counts for each element of interest were representative of the bulk powder. The relative concentrations were calculated according to the well-known k-ratio protocol, wherein corrections for atomic number, absorption, and secondary fluorescence are accounted (Reed and Ware, 1972; Lifshin et al., 1975; Fiori et al., 1976). The linearity of the EDS detector response as a function of elemental abundance (~0.1 to 16 wt.%) was verified by analyzing a series of NBS-certified standard reference materials (SRM 1193, 1194, and 1195, National Bureau of Standards), each one consisting of different absolute concentrations of the elements: Mn, P, S, Si, Cu, Cr, V, Mo, Ti, Al, and Zr in a balance of Fe. Further, three different areas were analyzed to ensure that the EDS spectrum was consistent for each sample. The overall error relative to SRM values for all elemental components was estimated to be ±8%.

For mineral analyses (bulk mineralogy and clay fraction), samples from each zone of the dike (n = 4) were disaggregated and analyzed using X-Ray Diffraction (XRD) with a Rigaku Ultima IV at Southern Illinois University. All samples (bulk and clay fractions) were analyzed with a Cu rotating anode (1.54 Å wavelength), 40 kV, 44 mA, a sampling interval of 0.02 degrees, and a scan speed of 0.6 s/step. Random powdered samples were analyzed for bulk mineralogy between 5 and 65° 2-theta.

Oriented clay fraction (< 2 μm) samples were analyzed between 2 and 35° 2-theta. To make the clay fraction analyses, powders from each zone were mixed with 100 mL of water and 10 mL of 5% solution tri-sodium phosphate to peptize the sample. This mixture was then run through an Eppendorf centrifuge at 1000 rpm for 5 minutes. The supernatant was reserved and centrifuged again at 4000 rpm for an additional 4 minutes. The remaining slurry was smeared onto a glass slide, air-dried, and analyzed.

Materials Data Inc. (MDI) Jade-2010 software was used to interpret XRD patterns and identify minerals present in samples for both bulk and clay fraction analyses. Data were treated conservatively; tentatively identified phases, potentially consistent with just one peak or peak shoulder, were not listed as present in Table 2.

Bulk uncut rock samples (n = 4) were analyzed by visible to near-infrared reflectance spectroscopy (VNR) using the hand held probe attached to an ASD TerraSpec Pro spectrometer at Southern Illinois University to obtain mineralogical observations comparable to orbital and lander VNR data for the Martian crust. This instrument measures wavelengths between 350 and 2500 nm. The resolution of the instrument is 3 nm at 700 nm and 6 nm at 1400/2100 nm. Samples were prepared by breaking open fresh surfaces. They were analyzed at room temperature, using a contact probe where the spectrometer was placed directly against the flattest surface. The resulting spectra were analyzed
using The Spectral Geologist (TSG™) software version 8 to identify minima at wavelengths that indicate specific minerals.

3. Results

3.1. Chemistry

Bulk chemistry has been measured and results are presented normalized to the darkest sample (presumed to be the least altered) to emphasize how the bulk chemistry changes with alteration (Table 1; Fig. 3). All three alteration zones are depleted in silicon, potassium, and manganese, and enriched in sulfur and calcium compared to the darkest sample. Sulfur increases while potassium and silicon decrease moving from the NW to the SE along the dike, consistent with an increase in alteration minerals. Both the purple and red samples are enriched in iron compared to the darkest sample. Aluminum, chromium, titanium, and phosphorous displayed no systematic changes and can be considered immobile in the system within the uncertainty of our measurements (note that for the red sample all values are slightly below one which is caused by the large sulfur, calcium, and magnesium influx and not necessarily by removal of these elements). This is consistent with fluid mobility removing Si and K but adding S, Fe, Ca, and possibly Mg presumably from the surrounding rock. However, the low-energy photons of Mg suffer much higher absorption compared to the high-energy photons of the heavier elements captured at the detector of the XRF EDS and therefore there is larger uncertainty in the measurement. We note, that the chemistry of the green dike is somewhat different from the other three, which we attribute to the larger amount of calcite in that sample compared with the rest (see the following Mineralogy section); thus, increasing the Ca content of the rock and decreasing several other elements (including iron) accordingly.

Fig. 1. Field site showing each of the four visually distinct alteration zones of the dike. The alteration and oxidation of the dike changes from the ‘darkest’ zone (a), the ‘green’ zone (b), the ‘purple’ zone (c), and the ‘red’ zone (d). The dike exhibits increasing oxidation from zone a through d. Hammer and chisel for scale. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).
3.2. Mineralogy

Mineralogy of all samples is consistent between measurement techniques (XRD and VNIR) with the differences being attributed to the sensitivity of the technique (Table 2). All samples contain phlogopite, which is from the igneous protolith. The only other igneous minerals found in all samples are sanidine-high and apatite. Even the least altered sample shows significant amounts of alteration minerals and while the crystal molds of former olivine are visible in hand sample (Fig. 2), olivine is not detected in any sample. Instead, all samples contain calcite, hematite, and kaolinite. Most samples also contain anatase, barite, halite, and dolomite. Kaolinite is present in all zones of the dike. Goethite, palygorskite and halloysite only formed in the most altered sample. The number of alteration phases increase in abundance with increasing alteration (Fig. 4; Table 1). The red sample has the most alteration phases including gypsum and the lowest abundance of primary igneous minerals.

4. Discussion

4.1. Conditions of alteration

Extensive previous work on thermochemical modeling of alteration mineralogy can provide constraints on the temperature, pH, and water to rock ratio during alteration (McAdam et al., 2008; Schwenzer and Kring, 2009; Filiberto and Schwenzer, 2013). Carbonates in our samples point to a near neutral pH and CO₂-bearing fluid (Bridges et al., 2019). Focusing on modeling results of neutral pH fluids can then be used to constrain temperature evolution of the system (e.g., Filiberto and Schwenzer, 2013). In order for Si to be mobile in a near-neutral pH fluid, high temperatures (> 200 °C) would be required (Filiberto and Schwenzer, 2013; Ruff and Farmer, 2016). Therefore, the system started at high temperatures with the magma intruding the sandstone causing a hydrothermal system at near-neutral pH and a dilute fluid.

Fig. 2. Hand samples with fresh cut surfaces of different regions of the dike: a) dark, b) green, c) purple, and d) red. All samples show some alteration but with increasing oxidation/alteration going from A to D. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).

Table 1

Bulk chemistry of each sample.

<table>
<thead>
<tr>
<th></th>
<th>Dark</th>
<th>Green</th>
<th>Purple</th>
<th>Red</th>
</tr>
</thead>
<tbody>
<tr>
<td>Na</td>
<td>N.D.</td>
<td>N.D.</td>
<td>1.06</td>
<td>1.63</td>
</tr>
<tr>
<td>Mg</td>
<td>3.91</td>
<td>1.32</td>
<td>3.26</td>
<td>6.19</td>
</tr>
<tr>
<td>Al</td>
<td>13.18</td>
<td>14.78</td>
<td>14.59</td>
<td>11.73</td>
</tr>
<tr>
<td>Si</td>
<td>27.54</td>
<td>25.76</td>
<td>21.02</td>
<td>19.36</td>
</tr>
<tr>
<td>P</td>
<td>1.91</td>
<td>2.34</td>
<td>1.66</td>
<td>1.56</td>
</tr>
<tr>
<td>S</td>
<td>0.68</td>
<td>1.01</td>
<td>0.87</td>
<td>2.80</td>
</tr>
<tr>
<td>K</td>
<td>6.17</td>
<td>5.74</td>
<td>2.61</td>
<td>2.52</td>
</tr>
<tr>
<td>Ca</td>
<td>14.03</td>
<td>34.43</td>
<td>14.37</td>
<td>19.60</td>
</tr>
<tr>
<td>Ti</td>
<td>5.77</td>
<td>7.45</td>
<td>5.17</td>
<td>4.53</td>
</tr>
<tr>
<td>Cr</td>
<td>0.28</td>
<td>0.36</td>
<td>0.23</td>
<td>0.23</td>
</tr>
<tr>
<td>Mn</td>
<td>0.36</td>
<td>0.33</td>
<td>0.29</td>
<td>0.23</td>
</tr>
<tr>
<td>Cl</td>
<td>N.D.</td>
<td>N.D.</td>
<td>1.78</td>
<td>1.02</td>
</tr>
<tr>
<td>Fe</td>
<td>26.17</td>
<td>6.50</td>
<td>33.11</td>
<td>28.62</td>
</tr>
<tr>
<td></td>
<td>100.00</td>
<td>100.00</td>
<td>100.00</td>
<td>100.00</td>
</tr>
</tbody>
</table>

N.D. Not detected.

3.2. Mineralogy

Mineralogy of all samples is consistent between measurement techniques (XRD and VNIR) with the differences being attributed to the sensitivity of the technique (Table 2). All samples contain phlogopite, which is from the igneous protolith. The only other igneous minerals found in all samples are sanidine-high and apatite. Even the least altered sample shows significant amounts of alteration minerals and...
before cooling to produce the rest of the mineral assemblage (Fig. 5). The presence of carbonates, kaolinite, and hematite suggests high water to rock ratios, temperatures \( \geq 150 \, ^\circ C \), and \( > 0 \) but \( < 0.5 \, \text{mol CO}_2 \) based on models for alteration at Gusev Crater, Mars (Filiberto and Schwenzer, 2013). Higher CO2 concentrations would produce more carbonates but destabilize hematite, and low CO2 concentrations would not produce carbonates. Goethite would be produced at lower temperatures but still with a high water to rock ratios as the hydrothermal system cools. Finally, the presence of the clay minerals halloysite and palygorskite in the most altered sample are indicative of the extended circulation of hot brines with high concentrations of Mg and Si (Birsoy, 2002), and gypsum and halite would only precipitate from the fluid at the very end from a more concentrated brine.

### 4.2. Applications to Gusev Crater

At Gusev Crater, MER Spirit analyzed a potential hot spring and/or hydrothermal deposit (Schmidt et al., 2008; Yen et al., 2008; Filiberto and Schwenzer, 2013; Ruff and Farmer, 2016). The mineralogy of the area in Gusev Crater is quite similar to what we have found at the Colorado Plateau (Table 1): Sulfate sands at Paso Robles, Dead Sea, Shredded, Arad, Tyrone, and Troy; hematite-rich outcrops between

---

**Table 2** Mineralogy of each sample by technique and grain size.

<table>
<thead>
<tr>
<th>Sample Technique</th>
<th>Mineralogy</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dark Bulk VNIR</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Dark Bulk XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Dark XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Green Bulk VNIR</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Green Bulk XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Green XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Purple Bulk VNIR</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Purple Bulk XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Purple XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Red Bulk VNIR</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Red Bulk XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
<tr>
<td>Red XRD</td>
<td>Sulfate sands&lt;br&gt;Barite&lt;br&gt;Hematite&lt;br&gt;Goethite&lt;br&gt;Gypsum&lt;br&gt;Halite&lt;br&gt;Halloysite&lt;br&gt;Palygorskite</td>
</tr>
</tbody>
</table>
Fig. 5. The fluid history of the dike/water interaction during/after dike emplacement within the sandstone. Note that this figure focuses on the reactions in the dike. (A) Intrusion of the dike into the sandstone producing a hydrothermal system circulating through the sandstone and penetrating the dike after cooling below its ductile temperature and fracture formation; (B) shows the system as the dike and fluids cool. Green arrows indicate element mobility. Red arrows represent hot fluids, while blue arrows represent cold fluids. Orange sub-box shows the alteration minerals forming within the dike: yellow squares represent phlogopite, white laths represent feldspar altering to kaolinite, brown rhombohedra represent olivine casts, light blue represents calcite, and black lines represent oxide minerals; (C) shows the dike system after erosion and how it appears today – colors correspond to coloration of the dike shown in Fig. 1. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).

4.3. Applications to Jezero Crater, NE Syrtis, and the ‘Midway’ point

While we await a Mars 2020 landing in Jezero Crater, the mineralogy analyzed in the orbital data at Jezero and the surrounding NE Syrtis region is consistent with the results presented here. Jezero Crater contains a delta with strong evidence for carbonates associated with olivine (Goudge et al., 2015) and hydrated minerals. There is a potential volcanic floor unit in contact with sedimentary rocks (Goudge et al., 2015), which is where Mars 2020 will presumably land. The floor unit, if it represent a lava flow, could have formed a hydrothermal system with the sedimentary rocks below if they contained hydrated minerals, ground water, or the cryosphere. If this system formed, it could have been potentially similar to the Robbers Roost Dike field site, with the caveat that the Robbers Roost dike is a vertical dike system vs. a potentially thin horizontal volcanic floor unit (Goudge et al., 2015).

Such magma driven hydrothermal systems are more common in plutonic or intrusive systems (e.g., Hochstein and Browne, 2000), lava flows on Earth can produce similar hydrothermal systems to plutonic systems by dehydrating the substrate material (e.g., Griffiths, 2000; Hochstein and Browne, 2000). Dehydration of the substrate material could lead to melting of that substrate unit, which would then potentially cause thermal erosion (Griffiths, 2000). Therefore, geometry can affect specifics of the system, but should not substantially affect the mineralogical, geochemical, and habitability potential applicability of our results (e.g., Hochstein and Browne, 2000). What is more likely to change the outcome of such a contact in producing a hydrothermal system is the temperature and thickness of the lava flow and the nature (consolidated vs unconsolidated, anhydrous vs carbonaceous vs hydrate) of the sediments the lava contacts (Griffiths, 2000). At this point, those remain unknowns until Mars 2020 Rover lands in Jezero Crater and explores this important contact.

The larger regional area (NE Syrtis and Midway point) contains spectral signatures of olivine, pyroxene, smectite clays, sulfates, and carbonates (Ehmann and Mustard, 2012; Goudge et al., 2015; Salvatore et al., 2018). The region has a proposed volcanic capping unit directly in contact with sulfate-bearing sediments (Goudge et al., 2015; Salvatore et al., 2018), where hydrothermal alteration from contact between these units should occur due to remobilization of S, H2O, and/ or CO2 and, if it does, would have produced a habitable environment similar to the Robbers Roost Dike field site. Therefore, this site is an ideal analogue environment for informing site exploration of the crater floor once the rover lands, as well as a potential extended mission to the crater rim or the ‘Midway’ point, which is proposed to have similar geology and mineralogy to the broader NE Syrtis region (Bramble et al., 2018; Mustard et al., 2018).

4.4. Habitability of fluids

If the requirements for life on Earth are any measure for potential life on Mars, a habitable environment requires liquid water, nutrients, an energy source, and protection from detrimental influences such as a cold, highly oxidizing conditions, and high flux of ionizing radiation (Conrad, 2014). Based on that assumption, the hydrothermal system formed during interaction of the magma (heat source) and groundwater would have been a habitable environment once the system cooled below ~120 °C (Rothschild and Mancinelli, 2001; Mancinelli et al., 2004; DasSarma, 2006). The fluid entering the system would have contained C, S, and Fe, which are all key bio-essential elements, and could potentially be used as an energy source for chemolithotrophic...
(Westall et al., 2015; Price et al., 2018) microorganisms (microorganisms that obtain their energy from the oxidation of reduced inorganic compounds), which could utilize the carbon to produce complex organic molecules from inorganic molecules (Cockell et al., 2016). Therefore, it is paramount that future missions look at the interface of sedimentary rocks with magmas (or impact melts), where microbial or cellular life, if present, could have taken advantage of a selection of favorable temperatures with guaranteed liquid water on a frozen planet. Moreover, active redox processes, geochemical gradients, and a moving water column to bring fresh fluid and take waste away, whilst being protected from the detrimental conditions at the Martian surface, would be an ideal site for future sample analyses. Further, such hydrothermally altered igneous samples should be targeted for sample caging by Mars 2020 as they represent a previously habitable environment that could preserve ancient Martian biosignatures, assuming there had been previous life.

CRediT authorship contribution statement

Lacey J. Costello: Writing - original draft, Formal analysis, Investigation, Visualization, Validation. Justin Filiberto: Conceptualization, Validation, Investigation, Data curation, Writing - review & editing, Supervision, Project administration, Funding acquisition. Jake R. Crandall: Conceptualization, Validation, Investigation, Writing - review & editing, Sally L. Potter-McIntyre: Conceptualization, Validation, Investigation, Data curation, Writing - review & editing, Funding acquisition. Susanne P. Schwenzer: Conceptualization, Investigation, Data curation, Writing - review & editing, Funding acquisition. Daniel R. Hummer: Validation, Investigation, Writing - review & editing, Formal analysis, Resources. Karen Olsson-Francis: Writing - review & editing. Scott Perl: Writing - review & editing.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgements


