Strength-ductility trade-off via SiC nanoparticle dispersion in A356 aluminium matrix


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Strength-ductility trade-off via SiC nanoparticle dispersion in A356 aluminium matrix

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Abstract

A process was developed to disperse β-SiC nanoparticles (NPs), with a high propensity to agglomerate, within a matrix of A356 aluminum alloy. A suitable dispersion of 1 wt.% SiC NPs in the A356 matrix was obtained through a hybrid process including a solid-state modification on the surface of the NPs, a two-step stirring process in the semi-solid and then the liquid-state, and a final hot-rolling process for fragmentation of the brittle eutectic silicon phase and porosity elimination. Titanium and nickel where used as the nanoparticle SiC surface modifiers. Both modifiers were found to improve the mechanical properties of the resulting material, however, the highest improvement was found from the nickel surface modification. For the nickel modification, compared to the non-reinforced rolled alloy, more than a 77 %, 85 %, and 70 % increase in ultimate tensile strength (UTS), yield strength (YS), and strain % at the break, respectively were found with respect to the unreinforced rolled A356. For the rolled nanocomposite containing 1 wt. % SiC_np and nickel modification, an average YS, UTS, and strain % at the break of 277MPa, 380 MPa, and 16.4% were obtained, respectively, which are unique and considerable property improvements for A356 alloy.

Keywords: SiC NPs; aluminium nanocomposite; nanoparticles; mechanical properties.
1- Introduction

The unabated thirst for fuel-saving and cost-efficient materials in automotive, defense, and aerospace industries has generated a focus on metal matrix nanocomposites (MMNCs). The primary goal of MMNC research is to employ nanoparticle (NP) reinforcements to produce high strength/low-density materials with a higher ductility than can be achieved by using micron-size reinforcements. In this case, a suitable dispersion of NPs can be used as a barrier for the movement of dislocations and avoid their pile up on grain boundaries [1]. However, the characteristic nature of the high surface-to-volume ratio and van der Waals forces between the NPs tend to keep them in the agglomerated state. Several different approaches have been examined (e.g. sonication, particle heat treatment, and various mixing methods) to break the bonds and thereby deagglomerate the clusters [2, 3].

Two major routes for MMNCs processing have been developed, solid-state processing (e.g. powder metallurgy) and liquid-state solidification processing [4]. In the powder metallurgy route, high shear stress generated by high-energy ball milling is used to shear nanoparticle clusters. However, this technique contains various drawbacks. The cost of metal powder production is high, and this method has limited capability for shaping products in geometrically complex forms. Furthermore, this method needs additional processes to manufacture the final parts, such as HIP’ing, machining, or another surface finishing [5, 6]. The casting method (liquid-state solidification processing) can be conducted in liquid, semi-solid, or in a two-step process, which has been used to overcome the noted drawbacks of the powder metallurgy route. This method is suitable to produce
components with complex shapes in any sizes. Also, it is a low cost, convenient, and accessible technique. The Casting of an MMNC where Orowan strengthening is dominant requires dispersion of NPs in the liquid melt followed by the engulfment of the particles within the solidifying grains which is one of the most challenging methods for achieving a uniform NP dispersion. It is known that a uniform distribution of NPs inside the grains is difficult to achieve in practice to pin the dislocations. Whether particles are pushed or engulfed during solidification depends on the velocity of the particles relative to the solidification front (SF) during grain growth. If the SF velocity is higher than the critical velocity, the particles will be engulfed by the moving SF. Assuming the velocity of the SF is held constant, NPs in low concentrations can thereby be engulfed [1]. Therefore, to design the microstructure of MMNCs, the interaction of NPs with an advancing solid-liquid interface and methods for their injection to the melt need to be studied [7].

Various measurements have been taken to address the problems associated with NPs agglomeration, such as semi-solid stir casting (SS-SC) [2-4], ultrasonic treatment [5-13], ball-milling [14-16] as well as a combination of these routes [17-20]. It has been reported that in the SS-SC process, shear stress in the two-phase region causes dis-agglomeration. Rohatgi [21] and his research team investigated the manufacture of aluminum-based nanocomposites by using a reactive wetting method in which magnesium chips were added to react with the alumina layer on the surface of aluminum alloy. They reported that magnesium, as it is introduced in the molten metal, could eliminate the oxygen and surface alumina by reacting with Al$_2$O$_3$. Indeed, Mg improves the wettability by removing surface alumina and decreasing surface tension or interfacial energy between particles and melt. They
showed that in the case of achieving an ideal distribution of nanoparticles without any clustering, the grains remain coarse and Orowan is the dominant strengthening mechanism. If agglomeration takes place, the clusters are moved by the solidification front and accumulation of clusters leads to the production of smaller grains, and the dominant strengthening mechanism, in this case, is Hall-Petch.

The idea of using a carrier protein, which is a method used in medical science to inject specific substances into the blood was suggested by Su et al. [22]. In this study, it was showed that aluminum powder is a suitable carrier to introduce nanoparticles into the melt. However, no low-magnification microstructure was reported in their study for concluding the method of incorporation. In 2014 Rohatgi et al. [23] evaluated strengthening mechanisms with the addition of magnesium as the wetting factor to the melt. Nano-alumina powders were added to melt in two forms, once covered in foils without compression, and once compacted and pressed by foils. However, the agglomeration of NPs was shown in their results. Zhou et al. [24] used nickel coatings for nanoparticles before adding them into the melt to improve the wettability. However, sub-micron sized clusters were detected inside the grains. Also, it was not reported in their study that how they coated nano-sized ceramics with a nickel layer, which individually separately is not a trivial task. Two studies were conducted by Rohatgi et al. [1, 25] regarding the Brownian motion of nanoparticle and the possibility of trapping these particles by the solidification front. They showed that the Brownian motion of nanoparticles or low rates of nanoparticles could be dominant, and there is a critical amount for nanoparticles to be added in composites. They illustrated that by the addition of nanoparticles in small volume fractions, they could be trapped and captured by solidification front due to Brownian motion, while larger submicron particles are
more prone to be moved by solidification front and clustering. Zhang et al. [20] milled nano silicon carbide powders with Al2014 alloy powders to improve their wettability within the liquid alloy. They reported an improvement in strength, while provided microstructures which showed agglomeration of NPs in the metal matrix. A distinct study was reported [12], in which a liquid Mg-Zn matrix was reinforced by SiC nanoparticles. They dispersed 1%. vol NPs in the liquid matrix by using ultrasonic waves, and a high percentage of matrix alloy was vaporized via an assisting vacuum pump. They achieved a unique and ideal microstructure. It should be noted that this method is not suitable and applicable to all types of light metals, as the matrix should be evaporated to some extent. Vishwanatha et al. [13] tried to provide an approach that would eliminate the disadvantages of both contact and non-contact ultrasonic methods for the dispersion of nanoparticles. For this purpose, a contact-type ultrasonic method was used. Then the liquid mixture was transferred into another chamber with a bottom pouring system to enhance the distribution of nanoparticles using a non-contact ultrasonic method. They provided the microstructures from three methods of 1. Contact, 2. non-contact and 3. A combination of contact and non-contact types, which showed the inability of contact and non-contact ultrasonic methods for the distribution of nano-ceramic particles in their metal matrix. It also was observed that while a better dispersion was obtained with the combination method, an appropriate uniform distribution was not achieved in the produced nanocomposites. The contamination of the liquid melt because of contact between the probe tip and the liquid melt was reported to be a severe problem and for the non-contact ultrasonic treatment, the acoustic energy to the liquid melt is transmitted through the mold wall and the drawback of this method is that the processing time is typically too short to achieve complete
deagglomeration. Also, the ultrasonic intensity is very limited on the surface of the melt, and it is a time-consuming process to introduce the nanoparticles into the melt using ultrasonic vibration.

To the best knowledge of authors, a uniform distribution of nanoparticles has not been accomplished in Al-Si alloys by the mentioned methods, and the reinforcements still have a relatively poor wettability and a high tendency to cluster propagated by the present Van-der-Waals forces.

A356 is a hypoeutectic Al-Si alloy, consisting of plate-like eutectic silicon and \( \alpha \)-aluminium primary phase dendrites. To improve the mechanical properties of these alloys, efforts have been made to develop aluminium matrix composites (AMCs) reinforced by ceramic particles. In Table 1, a summary of related reports for A356-based composites is tabulated. It can be observed that various methods and reinforcement have been applied to reinforce A356 alloy, and in most of them, a composite elongation below 10% was obtained. The most critical issue for these types of Al-Si alloys is the brittle plate-like eutectic silicon that causes failure during tensile loading even if an ideal dispersion of nanoparticles can be obtained. In fact, for the reinforcing of such alloys by NPs, eutectic silicon fragmentation into fine particles (or modification of eutectic silicon phase) is required in parallel to NPs addition.
Table 1
Mechanical properties reported for A356 composites in the literature.

<table>
<thead>
<tr>
<th>Materials</th>
<th>Reinforcement Size and Type</th>
<th>Manufacturing Method</th>
<th>YS (MPa)</th>
<th>UTS (MPa)</th>
<th>Break Strain %</th>
<th>Remarks</th>
<th>Ref</th>
</tr>
</thead>
<tbody>
<tr>
<td>A356-1 wt. % SiC</td>
<td>Multi-sized particles</td>
<td>SS-SC followed by hot extrusion</td>
<td>131/-</td>
<td>228/-</td>
<td>8.5/-</td>
<td>Metallic carrier was used for a better wettability</td>
<td>[26]</td>
</tr>
<tr>
<td>A356-1 wt. % SiC</td>
<td>40 nm, nanoparticles</td>
<td>SS-Squeeze casting followed by Ultrasonic assistance</td>
<td>-</td>
<td>181/-</td>
<td>2.7/-</td>
<td>After ultrasonic: UTS=269 Break Strain=6.2 Using Al powders as a modifier</td>
<td>[27]</td>
</tr>
<tr>
<td>A356-2 wt. % SiC</td>
<td>40 nm, nanoparticles</td>
<td>LS-SC followed by Ultrasonic assistance and T6</td>
<td>144/-</td>
<td>259/-</td>
<td>5.3/-</td>
<td>Using Al powders as a modifier</td>
<td>[28]</td>
</tr>
<tr>
<td>A356-3 wt% ZrB₂+0.05 wt%Er</td>
<td>10 to 80 nm, nanoparticles</td>
<td>LS-electromagnetic stirring</td>
<td>250/-</td>
<td>325/-</td>
<td>13.65/-</td>
<td>Using in situ method</td>
<td>[29]</td>
</tr>
<tr>
<td>A356-0.9 wt % CNFs</td>
<td>Diameter of 50 to 200 nm, and length of 3 to 12µm, nanowires</td>
<td>LS-ultrasonic vibration followed by extrusion</td>
<td>202/-</td>
<td>261/-</td>
<td>5/-</td>
<td></td>
<td>[30]</td>
</tr>
<tr>
<td>A356-0.5 wt.% MWCNT</td>
<td>Outside $\phi$ 20–40 nm, inside $\phi$ 5–10 nm and length 10–30 nm, nanotubes</td>
<td>LS-SC followed by T6</td>
<td>-</td>
<td>-/277</td>
<td>-</td>
<td></td>
<td>[31]</td>
</tr>
<tr>
<td>A356-1.5 wt.% MWCNT</td>
<td>Diameter of 10-100 nm, carbon nanotubes</td>
<td>SS- squeeze casting</td>
<td>157.7</td>
<td>245</td>
<td>6.93</td>
<td></td>
<td>[32]</td>
</tr>
<tr>
<td>A356-3 wt% ZrO₂</td>
<td>30 nm, nanoparticles</td>
<td>SS-SC followed by T6</td>
<td>-</td>
<td>165</td>
<td>4.7</td>
<td></td>
<td>[33]</td>
</tr>
<tr>
<td>A356-1 vol.% SiCp</td>
<td>50–56µm, particles</td>
<td>LS-SC with Ultrasonic assistance followed by T6</td>
<td>-/183</td>
<td>-/257</td>
<td>-/7.5</td>
<td>Using hybrid stirring</td>
<td>[34]</td>
</tr>
<tr>
<td>A356/5 vol.% SiCp</td>
<td>10 µm, particles</td>
<td>LS-SC</td>
<td>120/-</td>
<td>160/-</td>
<td>4/-</td>
<td>Using Al powders as a modifier</td>
<td>[35]</td>
</tr>
<tr>
<td>A356/1.5 vol.% Al₂O₃</td>
<td>20 nm, nanoparticles</td>
<td>LS-SC</td>
<td>230/-</td>
<td>265/-</td>
<td>1.12/-</td>
<td>Using Al powders as a modifier</td>
<td>[36]</td>
</tr>
<tr>
<td>A356/1.5 vol.%</td>
<td>20 nm, nanoparticles</td>
<td>LS-SC</td>
<td>277/-</td>
<td>364.9/-</td>
<td>2.6/-</td>
<td>Using Al powders as a modifier</td>
<td>[37]</td>
</tr>
<tr>
<td>Material</td>
<td>Particle Size</td>
<td>Modifier</td>
<td>Process</td>
<td>Notes</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>--------------------------</td>
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<td>--------------------------</td>
<td>--------------------------------------------</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356/1.5 vol.% Al₂O₃</td>
<td>20 nm, nanoparticles</td>
<td>LS-SC</td>
<td>115/- 182/- 2.2/-</td>
<td>Using Al powders as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-1 wt% SiC</td>
<td>35 nm, nanoparticles</td>
<td>LS-SC</td>
<td>137/- 173/- 5.38/-</td>
<td>Using Al powders as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-5 wt% of Ni-coated Al₂O₃</td>
<td>50 nm, nanoparticles</td>
<td>LS-SC</td>
<td>135.42/- 186.36/- 1.27/-</td>
<td>Electroless deposition Coating on nanoparticles</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-2 wt% of Al₂O₃</td>
<td>80 nm, nanoparticles</td>
<td>LS-SC</td>
<td>- 160/- 1/-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-5 wt% Al₂O₃</td>
<td>60 nm, nanoparticles</td>
<td>SS-SC followed by squeeze casting</td>
<td>245/- 282/- 8/-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-0.1 wt% TiC</td>
<td>70-80 nm</td>
<td>LS-SC followed by T6</td>
<td>181/197 267/284 6.8/8.3</td>
<td>Using T6 heat treatment to modify alloy</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-15 wt% SiC</td>
<td>25 µm, particles</td>
<td>Electromagnetic LS-SC</td>
<td>- 309.83/- 13.5/-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-2 wt% Al₂O₃</td>
<td>40 nm, nanoparticles</td>
<td>SS-SC</td>
<td>- 195/- 6.8/-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-2 wt% SiC</td>
<td>≤30 nm, nanoparticles</td>
<td>Ultrasonic assisted LS-SC</td>
<td>150/- 300/- 4/-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-5 vol.% SiCp</td>
<td>4 µm, particles</td>
<td>LS-SC followed by Rheo-process</td>
<td>- 268/- 5.2/-</td>
<td>Using a twin-screw machine to apply sufficient shear stress on particle clusters</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-1 vol.% SiC</td>
<td>40 nm, nanoparticles</td>
<td>SS-SC followed by Hot-rolling</td>
<td>196/- 292/- 12.4/-</td>
<td>Titanium was used as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-5 vol.% SiCp</td>
<td>80 nm, nanoparticles</td>
<td>LS-SC</td>
<td>- 190/- 4.5/-</td>
<td>Aluminium was used as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-2 vol.% SiCp</td>
<td>40 nm, nanoparticles</td>
<td>Ultrasonic assisted LS-SC followed by Hot-extrusion</td>
<td>- 226/- 5.5/-</td>
<td>Aluminium was used as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-3 wt% B₄C</td>
<td>100-300 nm, nanoparticles</td>
<td>LS-SC followed by squeeze cast</td>
<td>- 320/- -</td>
<td>Aluminium was used as a modifier</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-8.37 vol% TiB₂</td>
<td>In-situ particles</td>
<td>LS-SC followed by Hot-extrusion</td>
<td>217/347 258/391 2.73/1.32</td>
<td>In-situ method</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>A356-15 wt% SiC-5 wt% Fly-ash</td>
<td>25 µm, particles</td>
<td>Electromagnetic LS-SC followed by heat treatment</td>
<td>- 320/364 -12</td>
<td>Using heat treatment to modify alloy</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>
In this study, A356 based nanocomposites were prepared by solid-state modification of NPs using nickel and titanium as modifiers of the NPs, then stir casting process in two steps, followed by hot rolling (HR) to study the dispersion of β-SiC NPs within the matrix and associated physical effects.

2- Experimental procedures

2.1. Preparation of reinforcing phase

The reinforcement for the fabrication of the A356-based MMNCs was 1 wt% SiC NPs with an average particle size of 50 nm obtained from US-Research NanoMaterial Co. The benefit of the selection of SiC NPs is their higher thermal conductivity and modulus compared to Al₂O₃ NPs, for example. Besides, the relative
density of SiC is closer to that of aluminium compared with alumina. Nickel and titanium (Shanghai Dinghan Chemical Co., Ltd. China) as the modifier agents were used to improve the dispersion of nanoparticles in the A356 matrix. The reason for using them as the modifier for the deagglomeration of nanoparticles was reported elsewhere [26]. For this purpose, a Fritsch Pulverisette P5 planetary ball mill was used under high purity (99.99%) argon gas. Details of the ball-milling process are given in Table 2.

In the semi-solid metal, a high viscosity fluid can mechanically trap composite powders, including NPs in which case mechanical mixing can improve the mixing of composite powders in the composite. It should be mentioned that the mechanical shearing used for solidification processing in traditional metal matrix composites (with microparticles) does not work for submicron and nanosized particles, as the shearing force generated by the mechanical stirring decays rapidly and become too weak to separate the nanoparticles. This is a primary reason why the solid-state modification of NPs before powder-injection to the melt was examined in this study.

2.2. Preparation of the bulk composite

Bulk MMNCs were manufactured using an injection of the prepared composite powders into molten A356 aluminum alloy under high purity (99.999%) argon gas atmosphere. A356 alloy contains around 7 wt.% silicon that suppresses the potential chemical reaction between liquid aluminum and the SiC NPs, and it prevents the formation of $\text{Al}_4\text{C}_3$ (brittle compound) at the particle-matrix interface below 700 °C. After the entire alloy in the crucible was melted, it was cooled to 600 °C, corresponding to a 0.3 solid fraction, based on analysis with Thermo-calc
software [52]. At 600 °C, severe oxidation and burning of NPs can be avoided. Then, 1 wt. % Mg chips as a wetting agent, prepared by machining, were added and stirring of the semi-solid alloy (using a ceramic coated stainless-steel impeller) at 400 rpm was initiated, while pre-heated powders (pre-heating was done at 350 °C for 120 min in argon atmosphere to remove the moisture and impurities) along with Mg chips were added to the uniformly formed vortex over a period of approximately 6 min to increase the incorporation rate of SiC NPs within the liquid aluminum alloy. Stirring was stopped after 6 minutes of stirring in semi-solid condition, and the melt temperature was increased to 680 °C with a heating rate of 25 °C/min. It was observed previously [26, 59] that the semi-solid condition is not enough for the required interaction of Ti and Ni with the melt such that a suitable releasing of the NPs did not occur. Therefore, the melt was stirred at 680 °C for an extra 2 minutes after 6 minutes stirring at the semi-solid state for a higher release of the NPs. After the completion of the stir casting process, the bottom valve was opened, and the melt was guided to a mould at room temperature. The samples produced by two-step stir casting were then prepared for the HR process. Fig. 1 shows the schematic of the manufacturing process of bulk MMNCs, which was in this study.

**Table 2**

**Characteristics of the samples fabricated in this study.**

<table>
<thead>
<tr>
<th>Matrix</th>
<th>Reinforcement</th>
<th>Ball milling Time (h)</th>
<th>Ball milling Speed (rpm)</th>
<th>Ball to powder weight ratio</th>
<th>Semi-Solid Stirring Time (min)</th>
<th>Semi-Solid Stirring Tem (°C)</th>
<th>Liquid-state Stirring Time (min)</th>
<th>Liquid-state Stirring Tem (°C)</th>
<th>Hot-Rolling Tem (°C)</th>
</tr>
</thead>
<tbody>
<tr>
<td>A356 (480±5g)</td>
<td>Ti (15g)-SiC&lt;sub&gt;n&lt;/sub&gt; (5g)</td>
<td>24</td>
<td>220</td>
<td>8:1</td>
<td>6</td>
<td>600</td>
<td>2</td>
<td>680</td>
<td>500</td>
</tr>
<tr>
<td>A356 (480±5g)</td>
<td>Ni (15g)-SiC&lt;sub&gt;n&lt;/sub&gt; (5g)</td>
<td>24</td>
<td>220</td>
<td>8:1</td>
<td>6</td>
<td>600</td>
<td>2</td>
<td>680</td>
<td>500</td>
</tr>
</tbody>
</table>
Fig. 1. Schematic of the manufacturing process implemented for this study.
2.3. Material characterization

Microstructure characterizations were performed using a field emission scanning electron microscope (FE-SEM, Mira 3, Tescan), operating at 10 kV and additionally equipped with an energy-dispersive X-ray spectrometer (EDS). For SEM study, the samples were ground, polished, and etched with Keller's reagent (190 ml water, 5 ml HNO₃, 3 ml HCl, and 2 ml HF). Fractography was studied using a scanning electron microscope (SEM, Vega, Tescan, Electron Optic Services).

The phase composition, NP dispersion, and their interaction with dislocations were studied with a transmission electron microscope (TEM, JEM-2100, Jeol), operating at 200 kV. Samples for TEM analyses were prepared by site-specific Ga⁺ focused ion beam milling process (FIB, 450S, FEI).

The tensile tests were performed at room temperature using a uniaxial tensile universal testing machine (Model 5982, Instron) of 100kN operating capacity at a constant rate of crosshead displacement, with an initial strain rate of $2 \times 10^{-3} \text{ s}^{-1}$.

For this purpose, plate tensile specimens with an 8.6 mm gauge length were prepared by the ASTM E8M standard and three repetitions. The YS (yield strength), UTS (ultimate tensile strength), and ductility (percent elongation to break) were measured.

3- Results and discussion

As described, solid-state modification of NPs using Ti and Ni metallic powders that effectively caused NPs breaking up due to repeated collisions between balls and powders and then semi-solid stirring at low temperature has some advantages as a suitable combined method for NPs dispersion within the melt. Figs. 2 and 3 show the microstructure of nanocomposites containing Ti and Ni as the modifier agents,
respectively, indicating that following liquid-state stirring after semi-solid stirring is a suitable complementary process that can complete gradual releasing of NPs. The formation of isolated, binary, and ternary SiC NPs after modification and their mechanical bonding with Ti and Ni without any reaction occurrence [26, 59] could provide adequate wettability with the molten aluminium, and ensures their proper distribution by gradual releasing in the melt. Suitable dispersion of NPs within the grains with no observed segregation on grain boundaries can be observed (Figs. 2a, b and 3a, b) for the two samples; additionally, no Ni segregation on the grain boundaries (Fig. 3c) was detected. Detection of many binary and ternary (or more) NPs in as-cast microstructure after their solid-state modification indicated that even intensive mechanical collisions of heavy balls cannot completely deagglomerate them, while it seems that using of a suitable metallic modifier followed by semi-solid and then liquid-state stirring can be an appropriate method for MMNCs manufacturing with uniform reinforcement distribution.
Fig. 2. SEM-SE micrograph of surface morphology of as cast samples for (a, b) Ti-1 wt. % SiC<sub>n</sub>.
Fig. 3. SEM-SE micrograph of surface morphology of as cast samples for (a, b) Ni-1 wt. % SiC\textsubscript{np}, (c) SEM-SE micrograph of surface morphology of as cast Ni-1 wt. % SiC\textsubscript{np}, and corresponding EDS elemental mapping for Al, Ni and Si elements.

Fig. 4 shows a schematic of processes that occurred for the samples during stirring and solidification. It has been observed that the final microstructure of the developed A356 alloy consists of lamellar and flake-like Si plates embedded in the α-Al matrix without any tendency for segregation along the grain boundaries. Also, it can be observed that the interaction of Ti and Ni with aluminium melt (exothermic reaction) caused gradual releasing of NPs, resulting in their final engulfment and dispersion inside the grains.
Fig. 4. Schematic of the microstructures obtained after solidification in this study.

Since Brownian motion can be considered as a mechanism that retards the movement of NPs, it seems that it will enhance the capture of NPs by the solidification front and improve their dispersion, therefore improving the chances of achieving Orowan strengthening. According to the Orowan mechanism, dislocation bowing is necessary for dislocations to bypass NPs. This motion of fine
NPs can produce a considerable random movement that interrupts the settling behaviour of a particle and it has been reported that this motion will be considerable by decreasing the average size of NPs below 100 nm [25].

After casting, the HR process was applied to remove porosity, and related microstructure is shown in Figs. 2a and 3a. Also, this process was applied to change the morphology and shape of flake-like eutectic silicon and make fine fragmented silicon particles. Figs. 5 and 6 show SEM micrographs after the HR process, where we can observe how fine Si particles between 1-4 µm were obtained after the HR process. Also, a more suitable dispersion of fine and coarse NPs with a lower interparticle distance was obtained for the sample in which Ni was used as the modifier. To the best knowledge of authors, this suitable dispersion of NPs shown in Fig. 6a is among the best ones in aluminium matrix nanocomposites. It should be noted that the inter-particle distance of an ideal MMNCs reinforced by 1 wt. % NPs should be considerably lower than as obtained in this study, indicating that the manufacturing of perfect distribution of nanoparticles in the MMNCs is complicated.
Fig. 5. SEM-SE micrograph of rolled Ti-1 Wt. % SiC<sub>np</sub>, showing (a) remains of larger Si particles, and (b) fine and coarse NPs dispersion in matrix.
Fig. 6. SEM-SE micrograph of rolled Ni-1 wt. % SiC<sub>np</sub> (a) low, (b) high magnification.
Fig. 7 shows electron-transparent FIB-produced lamella, containing Si fragments and almost an ideal dispersion of SiC NPs, without any agglomerates.
Fig. 8. BF-TEM micrograph of Ni-1 Wt. % SiCnp 2s sample showing interaction of dislocations with fine SiC NPs at various areas.

The interaction between dislocations, controlling the mechanical properties of the investigated material, and the finely dispersed nanoparticles, are presented in Fig. 8, showing results obtained in three different areas. Several important conclusions can be drawn from these nanostructures: first, dislocation interact with NPs, and bowing mechanism occurred for fine SiC NPs, while no interaction of dislocations with coarse NPs was observed; Second, in some parts dislocation loops were observed; Third, the interaction of fine NPs with dislocation in this area was considerable. From Fig. 7 we can conclude that this investigated area is close to grain boundary as much silicon fragmentation was present. In this aspect, the NPs
and dislocation interaction, combined with the bowing of dislocations, suggests that the presence of fine NPs can reduce and affect dislocation pile-up at grain boundaries. It should be mentioned that a negligible grain refinement was observed after casting of nanocomposites (see supplementary Figs. S1, S2, and S3) and based on this interaction that was shown in Fig. 8, it can be deduced that the Orowan mechanism could be the main strengthening mechanism for the produced samples.

Fig. 9. a) TEM micrograph of a fine NP within aluminium matrix for Ni-1 Wt. % SiC$_{np}$ 2s sample, identified as $\beta$-SiC, and (b) corresponding EDS spectra.

Based on the study by Xu et al. [60], more NP capture during solidification can occur when a core-shell nanoparticle, including a metallic core and ceramic shell face (SF). Following this report, some of the researchers in previous studies [59, 61, 62] tried to show the effect of metallic carriers (modifiers) on the Hamaker constant of ceramic NPs during solidification in aluminium matrix nanocomposites. Fig. 9 presents a TEM micrograph of an individual SiC NP embedded in an aluminium matrix as well as results of corresponding EDS analysis, excluding Ni presence around NP, suggesting core-shell structure after ball-milling might disappear during
stirring and solidification due to a high interaction tendency of nickel and exothermic reaction with aluminium A356 melt at 680 °C [59]. Therefore, it seems that pure NPs facing the SF and Brownian motion might be the important reason for the capturing of NPs.

As mentioned for Table 1, many researchers tried to improve the mechanical properties (strength-ductility trade-off) of A356 aluminium alloy. For this purpose, it was found that the morphology of eutectic silicon is the main prerequisite for strength-ductility trade-off. A comparison between microstructures of MMNCs after HR and before HR (compare Figs. 5a and 6a with Figs. 2a and 3a) show that the HR process as a severe plastic deformation technique completely fragmented the eutectic silicon phase into fine silicon particles and disperse them in the matrix. After this modification, the nano-scale reinforcing of the matrix based on the Orowan mechanism can be the second step for improving the mechanical properties. For this purpose, detachment of NPs from their initial condition and dispersing them in a matrix of Ni and Ti in solid-state was performed in this study. The gradual releasing of the NP from the Ni and Ti modifiers during stirring was discussed, and considerable interaction of fine NPs with dislocations was observed.

Table 3, as well as Figs. 10 and 11 summarize the tensile behavior of the reinforced and unreinforced samples and update of strength-ductility trade-off dilemma for A356 aluminium alloy. Unreinforced alloy after HR has a considerable combination of ductility and strength compared to many reported values in Table 1 due to fragmentation of eutectic silicon. After composite manufacturing, three changes occurred for A356 alloy. First, the addition of 1 wt. % Mg as the wetting agent. Second, addition of 3 wt. % Ni and Ti (separately), and third, suitable dispersion of SiC NPs inside the grains. Based on the literature [63, 64], the addition of Mg, Ti,
and Ni is not negligible on the strength of A356 alloy, especially after the T6 process. However, more than 77% and 70% increase in MMNCs for the UTS and elongation to failure, respectively, by using Ni as the modifier of NPs is emanated from well dispersion of NPs and their interaction with dislocations for avoiding dislocation pile-up, supporting Orowan strengthening mechanism. From Figs. 5a and 2b, it can be deduced that the number of binary, ternary, and clusters of NPs in Ti-rich nanocomposite is higher than that of the other nanocomposite. It was reported in the literature [65] that at a temperature above 650 °C Ti, due to its tendency for oxidation, has a lower interaction with aluminum melt compared to Ni. Therefore, better performance of Ni as the modifier would be expected in the second step of stirring at liquid-state. It should be noted that some sparks were observed for Ni-rich samples during stirring at the second step of stirring at 680 °C. It is believed that more interaction of modifiers with the melt will lead to more heat release. Therefore, a higher amount of NPs releasing can occur, affecting the performance of the modifier. It was obtained that the Ti-rich sample had a considerable improvement in mechanical properties compared with the unreinforced sample, while its UTS and elongation to failure were lower than the Ni-rich samples.

Very recently, researchers [66, 67] reported strength-ductility trade-off for A356 aluminum alloy. Some impressive results were obtained in their study. Zhang et al. [65] used a complicated multi-step process, including four passes of equal channel angular pressing (ECAP) process with intermediate heat-treatment for the first step. Then after the T6 process, 80% cryo-rolling process was applied, and more than 480 MPa for UTS and 8.1% elongation to failure were obtained. Dang et al. [66] applied rapid solidification (96K/s) followed by post-heat treatment process
and subsequent solid-solution treatment and quenching and finally, artificial aging for obtaining nano-sized eutectic silicon particles. More than 350 MPa for UTS and 21% elongation to failure were obtained after these steps, indicating the importance of Si particle size and dispersion in the matrix. The interaction of dislocations with nano-sized silicon particles was shown in their study. However, Table 3 and Fig. 10 show that without any heat-treatment process, a considerable combination of UTS and elongation to failure was obtained in our study, indicating that if SiC NPs with a much higher Young modulus than Si is suitably dispersed in the matrix, then without any heat-treatment process, a trade-off of strength and ductility can be obtained. This is caused by the ceramic NPs acting as the barrier of dislocation movement. In fact, by simultaneous improvement of strength (by avoiding easy glide of dislocations), and ductility (by avoiding dislocation pile-up and stress accumulation at grain boundaries) the traditional concept of strength improvement that normally will lead to a reduction in elongation to failure will be overcome.

Table 3

<table>
<thead>
<tr>
<th>Samples</th>
<th>UTS (MPa)</th>
<th>Break Strain (%)</th>
<th>YS (MPa)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Unreinforced A356</td>
<td>214±6</td>
<td>9.6±0.1</td>
<td>149±2</td>
</tr>
<tr>
<td>3wt. % Ti-1wt. % SiCₙp</td>
<td>293±8</td>
<td>12.8±0.7</td>
<td>204±6</td>
</tr>
<tr>
<td>3wt. % Ni-1wt. % SiCₙp</td>
<td>380±3</td>
<td>16.4±0.1</td>
<td>277±5</td>
</tr>
</tbody>
</table>
Fig. 10. Stress-strain curve of prepared samples after the HR process.

Fig. 11. Comparisons between the present tensile properties of A356 aluminium alloys fabricated using alloying, grain refining, composites, ECAP process [66], rolling process [46, 66] and special casting process [29, 43, 66].
Fig. 12. The SEM-SE micrographs of the fracture surface of (a) Ti-1 Wt. % SiC$_{np}$, (b) Ni-1 Wt. % SiC$_{np}$ samples.

SEM micrographs of the fracture surface of the nanocomposite samples are presented in Fig. 12. More equiaxed dimples with a higher density were obtained
for Ni-rich samples compared with Ti-rich nanocomposite. Moreover, smaller and shallower dimples were revealed for Ti-rich nanocomposite. Silicon and NP cluster debonding and fracture resulted in lower matrix deformation and formation of more small and shallow dimples.

In the future, we will optimize the ball-milling process for modification of NPs to reduce the number of binary and ternary NP clusters. Using a lower average size for NPs could likely lead to even higher interaction with dislocations, improving the mechanical properties even further.

4-Conclusions

In this study, solid-state modification of as-received NPs was implemented utilizing ball-milling with Ni and Ti as surface modifiers. The composite powders with a matrix of Ni and Ti (separately) were added to the semi-solid melt of A356 aluminium alloy. The stirring of the slurry was conducted in the semi-solid melt, followed by a stop for heating the slurry, and then in liquid-state stirring was continued. Finally nanocomposite sheets were obtained by HR process. From the obtained results, the following conclusions can be drawn:

1- A suitable dispersion of fine and coarse NPs inside the grains was obtained after solidification and HR, especially for Ni-rich samples.
2- This suitable dispersion caused a considerable interaction of fine NPs with dislocations, and dislocation bowing was detected and shown.

3- With improved microstructure, a suitable combination of strength and ductility (380 MPa for UTS and 16.4 % elongation to failure) was obtained for the Ni-rich samples compared to previous reports.

4- No Ni was detected around an individual fine SiC NP showing that Ni modifier interacted with melt during stirring, and they were not effective on the solidification process of NPs.

5-Conflicts of Interest:

The authors declare no conflict of interest.

References


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