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## Investigating the water contents and hydrogen isotopic compositions of lunar apatite

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The initial studies of Apollo rock samples concluded that the Moon was an anhydrous planetary body [1]. Recently there has been a change of opinion with several research groups detecting copious amounts of water, up to 6050 ppm from a range of lunar glasses, melt inclusions and apatite [2-7]. Here we report results from insitu ion microprobe analysis of apatite grains from Apollo mare basalt 12064 and lunar basaltic meteorite Miller Range MIL 05035.

The  $\delta D_{SMOW}$  values for 3 apatite grains in 12064 range from +822 to +998 ‰, with the hydroxyl content ranging from ~2800 to 4744 ppm. Five apatite grains were analysed in MIL 05035 with  $\delta D$  values ranging from -137 to +830 ‰. Hydroxyl content ranges from ~470 to 2690 ppm. The analytical setup for MIL 05035 was such that it yielded relatively low count rates for <sup>2</sup>H (an order of magnitude less than those obtained for 12064), and this in turn introduced large errors into the analysis.

The results for ilmenite basalt 12064 are among the highest  $\delta D$  measured so far from lunar apatites [7] being more tightly constrained in both  $\delta D$  and water content. The results plot well outside the range for terrestrial water (+100 ‰ to -500 ‰) [8, 9], suggesting that the hydrogen isotopic composition measured in these apatites has not been compromised by terrestrial contamination. Cometary material has a high  $\delta D$  and therefore, it could be interpreted that at least some lunar hydrogen is of cometary origin [7-9]. The roles of other processes which potentially contribute to elevated D/H ratio in lunar apatites need to be further investigated.

We recorded  $\delta D$  values overlapping with terrestrial  $\delta D$  values for MIL 05035. However, it is unclear if any of the observed isotopic variation relates to the low, and variable hydroxyl content in this rock.

[1] Papike, J.J. (1998) *Min. Soc. Of America* [2] Saal, A.E. et al. (2008) *Nature.*, 454, 192-195 [3] Hauri, E.H. et al. (2011) *Scienceexpress.*, 333, 1-4 [4] Boyce, J.W. et al. (2010) *Nature.*, 466, 466-469 [5] McCubbin, F.M. et al. (2010) *PNAS.*, 107, 11223-11228 [6] McCubbin, F.M. et al. (2011) *GCA.*, 75, 5073-5093 [7] Greenwood, J.P. et al. (2011) *Nat. Geoscience.*, 4, 79-82 [8] Meier, R. & Owen, T.C. (1999) *Space Sci. Rev.*, 90, 33-43 [9] McKeegan, K.D. & Leshin, L.A. (2001) *Min. Soc. of America*.